# B.Tech. (Common for all Branches) 2nd Semester

# F. Scheme Examination, May-2015

## **ENGINEERING CHEMISTRY**

## Paper-CH-101-F

Time allowed: 3 hours]

[Maximum marks: 100

Note: (i) Question No. 1 is compulsory.

- (ii) Attempt four questions from remaining four sections selecting one question from each section.
- (iii) Use of non programmable calculator is allowed.
- 1. (a) Define the system having incongruent melting.

 $2 \times 10 = 20$ 

- (b) What are homogeneous and heterogeneous catalysis?
- (c) Differentiate triple point and eutectic point.
- (d) Define Break-point chlorination.
- (e) Define demineralization of water.
- (f) Describe stress cracking.
- (g) What do you understand by tinning?
- (h) Describe Iodine value of a lubricant.
- (i) Write uses of PF.
- (j) What do you understand by Bathochromic shift?

**24005-**P-4-Q-9(15)

[P.T.O.

10

## Section-A

2.	(a)	Draw and explain the phase diagram	of Zn-Mg
		system.	10

- (b) Write an explanatory note on Enzymatic catalysis.
- 3. (a) Draw and explain the phase diagram of H<sub>2</sub>O-system.
  - (b) Explain the concepts of promoters, inhibitors and poisoners. 10

#### Section-B

- 4. (a) 100 ml of water sample requires 20 ml N/50  $H_2SO_4$  during titration by using phenolphthalein indicator and 26 ml of same acid by using methyl orange indicator. Calculate the alkalinity of each type in terms of CaCO<sub>3</sub> equivalent.
  - (b) What do you understand by demineralization of water? Discuss in detail the ion-exchange process for demineralization of hard water with help of neat, clean and labeled diagram.
- 5. (a) A zeolite softener was 75% exhausted by removing the hardness completely when the 100000 litres of hard water sample passed through it. The exhausted zeolite bed requires 145 litres of 25% NaCl solution for its complete regeneration. Calculate the hardness of water.

24005

(b) Write	short	notes	on	
-----------	-------	-------	----	--

- (i) Caustic embritllement
- (ii) Boiler corrosion.

5×2

#### Section-C

- **6.** (a) Write short notes on:
  - (i) Role of Proper Designing in corrosion control.
  - (ii) Role of sacrificial anode in corrosion control. 5×2
  - (b) Write short notes on:
    - (i) Molybdenum disulphide as solid lubricant.
    - (ii) Semi-solid lubricants.

5×2

- 7. (a) Why additives are used in lubricants? Give some examples of additives, which are commonly used in lubricants.
  - (b) Write short notes on:
    - (i) Soil corrosion
    - (ii) Pitting corrosion.

5×2

## Section-D

- **8.** (a) Discuss the principle and application of DTA. 10
  - (b) Write short notes on:
    - (i) Differentiate thermosetting and thermoplastics.
    - (ii) Buna-N.

 $5\times2$ 

24005

[P.T.O.

http://www.haryanapapers.com

- 9. (a) Write the applications of U.V. and I.R. spectroscopy.
  - (b) Write short notes on:
    - (i) Ziegler-Natta Catalyst
    - (ii) Urea-formaldehyde resin.

5×2

24005